

Free electron theory of metals

- Metals are good conductors (both electrical and thermal)
- Electronic heat capacity has an additional (temperature dependent) contribution from the electrons.
- Why are some materials metals and others not?

Simple approximation: treat electrons as free to move within the crystal

Free electron theory of metals

- Alkali metals (K, Na, Rb) and Noble metals $Cu, Ag,$ Au) have filled shell $+1$ outer s-electron.
- Atomic s-electrons are *delocalised* due to overlap of outer orbits.
- Crystal looks like positive ion cores of charge +e embedded in a sea of conduction electrons
- Conduction electrons can interact with each other and ion cores but these interactions are weak because:

(1) Periodic crystal potential (ion cores) is orthogonal to conduction electrons - they are eigenstates of total Hamiltonian e.g. for Na conduct. electrons are 3s states, but cores are n=1 and n=2 atomic orbitals.

(2) Electron-electron scattering is suppressed by Pauli exclusion principle.

Assumptions:

- (i) ions are static adiabatic approx.
- (ii) electrons are independent do not interact.
- (iii) model interactions with ion cores by using an "effective mass" $m*$
- (iv) free electrons so we usually put $m^* = m_e$

Free Electron Model

Put free electrons into a very wide potential well the same size as the crystal i.e. they are 'de-localised'

Free electron properties

$$
i\hbar \frac{\partial \psi}{\partial x} = \pm \hbar k \psi \qquad -\frac{\hbar^2}{2m} \frac{\partial^2 \psi}{\partial x^2} = \frac{\hbar^2 k^2}{2m} \psi \qquad \frac{\partial \omega}{\partial k} = \frac{1}{\hbar} \frac{\partial E}{\partial k} = \frac{\hbar k}{m}
$$

Add a second piece of crystal the same size: The properties must be the same.

Density of states

Density of allowed states in reciprocal (k-) space is:

$$
\frac{\Delta K}{\delta k} \text{ in } 1 - D \qquad \text{or} \qquad \frac{\Delta V_K}{\delta k^3} \text{ in } 3 - D \qquad \text{x 2 for spin states}
$$

Fermi Energy

Fermi Energy, E_F

$$
Metals - HT10 - RJ \text{ Nicholas} \tag{5}
$$

Typical value for E_F e.g. Sodium (monatomic)

crystal structure: b.c.c. crystal basis: single Na atom lattice points per conventional (cubic) unit cell: 2 conduction electrons per unit cell 2 \therefore electrons per lattice point = 1

lattice constant (cube side) = $a = 0.423$ nm \therefore density of electrons $n = N/V = 2/a^3 = 2.6 \times 10^{28}$ m⁻³ ∴ E_F = 3.2 eV

Fermi Temperature T_F ? $k_B T_F = E_F$ ∴ $T_F = 24,000$ K

Finite Temperatures and Heat Capacity

Fermi-Dirac distribution function $f_{F-D} = 1/(e^{E-\mu/k_B T} + 1)$ electrons are excited by an energy $\sim k_BT$ Number of electrons is $\approx k_BT g(E_F)$ $\therefore \Delta E \approx k_B^2 T^2 g(E_F)$ ∴ $C_V = \Delta E / \Delta T \approx 2k_B^2 T g(E_F)$

$$
\begin{array}{ll}\n\text{Previously} & \therefore \ln \ n = \frac{3}{2} \ln \ E + \text{const.} \\
\text{we have } n = A E_{F}^{3/2} & \therefore \frac{dn}{n} = \frac{3}{2} \frac{dE}{E} \\
\frac{dn}{dE} = \frac{3}{2} \frac{n}{E_{F}} = g(E_{F}) \\
\therefore C_{v} = 3 \text{nk}_{B} \frac{k_{B}T}{E_{F}} = 3 \text{nk}_{B} \frac{T}{T_{F}}\n\end{array}
$$

∴ Heat Capacity is:

- (i) less than classical value by factor $-k_B T/E_F$
- (ii) proportional to $g(E_F)$

Is this significant?

Rigorous derivation

Magnetic susceptibility

• Susceptibility for a spin $\frac{1}{2}$ particle is:

$$
\chi = \frac{\mu_B^2 \mu_0}{kT}
$$
 / electron

• This is much bigger than is found experimentally - Why?

Pauli paramagnetism

Separate density of states for spin up and spin down, shifted in energy by $\pm \frac{1}{2}g\mu_B B$ (g=2)

Imbalance of electron moments Δn

 Δn = $\frac{1}{2}$ g(ε _F) × 2 μ _BB

giving a magnetization M

$$
M = \mu_B \Delta n = \mu_B^2 g(\epsilon_F) B
$$

and a susceptibility

$$
\chi = M/H = \mu_0 \mu_B^2 g(\epsilon_F) = 3n\mu_0 \mu_B^2 / 2\epsilon_F
$$

k-space picture and the Fermi Surface

T=0 states filled up to E_F
$$
\therefore \frac{\hbar^2 k^2}{2m} = E_F
$$

Map of filled states in k-space = Fermi surface 2 2 \hbar *F* $\therefore k_F = \sqrt{\frac{2mE}{\hbar^2}}$

or we can write:

$$
N = 2 \times \frac{4\pi}{3} \frac{k_F^3}{\left(\frac{2\pi}{L}\right)^3}
$$

$$
\therefore k_F^3 = \frac{3\pi^2 N}{V}
$$

k-space picture and the Fermi Surface

How big is Fermi surface/sphere compared to Brillouin Zone?

Simple cubic structure

volume of Brillouin Zone = $(2\pi/a)^3$

electron density $n = 1/a³$

volume of Fermi sphere = $4\pi k_F^3/3 = 4\pi^3/a^3$

= half of one B.Z.

Electron Transport - Electrical Conductivity

Equation of motion: Force = rate of change of momentum

$$
\hbar \frac{\partial k}{\partial t} = -e \left(\mathbf{E} + \mathbf{B} \times \mathbf{v} \right)
$$

Apply electric field - electrons are accelerated to a steady state with a drift velocity v_d - momentum is lost by scattering with an average momentum relaxation time τ

$$
\therefore \text{ momentum loss} = \frac{mv_d}{\tau} = -eE
$$
\n
$$
\therefore \text{ current } j = nev_d = \frac{ne^2\tau}{m}E
$$
\n
$$
\therefore \text{ conductivity } \sigma = \frac{ne^2\tau}{m} = ne\mu \qquad v_d = \mu E
$$

What happens in k-space?

All electrons in k-space are accelerated by electric field:

 $\hbar \delta k = F \delta t = - eE \delta t$

On average all electrons shifted by: $\delta k = -\frac{eE\tau}{h}$ \hbar

Fermi sphere is shifted in k-space by $\delta k \ll k_F$

∴ To relax electron momentum k must be changed by \sim k_F

Scattering occurs at E_F

∴ we need phonons with large value of k. But phonon energy is small so only a small fraction of electrons k_BT/ϵ_F can be scattered

Scattering processes

Basic Principle: Scattering occurs because of deviations from perfect crystal arrangement

Electron scattering mechanisms:

- (i) thermal vibrations i.e. phonons (vibrations of the atoms are a deviation from perfect crystal structure)
- (ii) presence of impurities charged impurities are very important - scattering is by Coulomb force i.e. Rutherford scattering.

Matthiessen's rule: Scattering rates $(1/\tau)$ add

$$
\therefore \rho = \frac{m}{ne^2} \sum \frac{1}{\tau} = \rho_T + \rho_i
$$

Mean free path (λ) :

electrons are moving with Fermi velocity v_F

 \therefore $\lambda = v_F \tau \ (NOT \ v_d \tau)$

Low temperature mean free paths can be very long as electrons are only scattered by impurities

Hall Effect

- In a magnetic Field **B** the electron experiences a force perpendicular to its velocity.
- A current **j** causes a build up of charge at the edges which generates an Electric field **E** which balances the Lorentz force

$$
(-e) \left(\mathbf{E} + \mathbf{v}_d \times \mathbf{B} \right)_y = 0; \quad E_y = (v_d)_x B_z
$$

Hall Effect $\left\| \begin{matrix} \text{In a magnetic Field } \mathbf{B} \\ \text{measurable for a perpendicular} \end{matrix} \right\|$ experiences a force perpendicular to its velocity.

A current **j** causes a build up of charge at the edges which generates an Electric Balance of forces: \parallel field **E** which balances the Lorentz force

$$
(-e) \left(\mathbf{E} + \mathbf{v}_d \times \mathbf{B} \right)_y = 0; \quad E_y = (v_d)_x B_z
$$

The Hall coefficient $\rm R_H$ is:

$$
R_H = \frac{E_y}{j_x B_z}
$$

$$
j_x = n(-e) v_d \Rightarrow R_H = -\frac{1}{ne}
$$

Negative sign is sign of the charge on the electron

Thermal conductivity

In metals heat is mainly carried by the electrons Simple kinetic theory formula for thermal conductivity K:

$$
K = \frac{1}{3}C\lambda v_F
$$
\n
$$
[C = \pi^2/3 k_B^2 T g(E_F) = \pi^2/2 n k_B k_B T/E_F]
$$
\n
$$
= \pi^2/6 \lambda v_F n k_B k_B T/E_F
$$
\n
$$
[\lambda = v_F \tau ; E_F = \frac{1}{2} m v_F^2]
$$
\n
$$
= \pi^2/3 m n k_B^2 \tau T
$$

Scattering processes

- Low temperatures: defects, τ independent of T
- Intermediate temp. : Low Temp phonons Debye model $\tau \propto T^{-3}$
- High temperatures: 'classical' phonons $\tau \propto T^{-1}$

Wiedeman-Franz ratio

Electrical and Thermal conductivities of electrons are both proportional to the relaxation time τ

Taking the ratio of the two should make this cancel so if we define the Lorenz number as $L = K/(\sigma T)$ we have the

Wiedeman-Franz Law:
$$
L = \frac{K}{\sigma} \frac{\pi^2 k_B^2}{3e^2}
$$

Predicted value is absolute and the same for all metals. Works well at high and low temps, - breaks down in 'Debye' region where energy and charge scattering are different

Successes and Failures of Free Electron Model

Successes:

- Temperature dependence of Heat Capacity
- paramagnetic (Pauli) susceptibility
- Ratio of thermal and electrical conductivities (Lorentz number)
- Magnitudes of heat capacities and Hall effect in simple metals

Failures:

- Heat capacities and Hall effect of many metals are wrong
- Hall effect can be **positive**
- Does not explain why mean free paths can be so long
- Does not explain why some materials are metals, some insulators and some are semiconductors

Nearly Free Electron Approximation

Use a travelling wavefunction for an electron, e^{ikx} , with kinetic energy $\hbar^2 k^2/2m$

- Assume that this is Bragg scattered by the wavevector $G=2\pi/a$ to give a second wave e^{i(k-G)x} with energy \hbar^2 (k-G)²/2m
- Crystal potential is periodic in real space. Therefore we can Fourier Transform the potential so that:

$$
V(x) = \sum_{G} V_{G} \exp(iGx)
$$

For a schematic solution we calculate what happens for a single Fourier component V_G so $V(x) = V_G(e^{iGx} + e^{-iGx})$

Nearly Free Electron Approximation

Use a travelling wavefunction for an electron, e ikx, with kinetic

Formally what we are doing is to solve the Hamiltonian form of Schrödinger equation

$$
H \psi = E \psi
$$

where ψ are the two travelling wave solutions. Expanding gives:

$$
\begin{pmatrix}\nH_{11} - \lambda & H_{12} \\
H_{21} & H_{22} - \lambda\n\end{pmatrix}\n\begin{pmatrix}\ne^{ikx} \\
e^{i(k-G)x}\n\end{pmatrix} = (E - \lambda)\n\begin{pmatrix}\ne^{ikx} \\
e^{i(k-G)x}\n\end{pmatrix}
$$
\n
$$
H_{11} = \left\langle \psi_1^* \middle| -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} \middle| \psi_1 \right\rangle = \frac{\hbar^2 k^2}{2m}, \quad H_{22} = \frac{\hbar^2 (k - G)^2}{2m}
$$
\n
$$
H_{12} = \left\langle \psi_1^* \middle| V(x) \middle| \psi_2 \right\rangle = V_G
$$